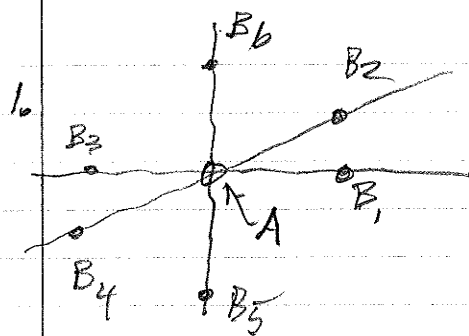


PHY 240A MidTerm #2 Solutions



$t_{AB} = t$ A-B hopping

(It would be simple to add B-B near neighbor hopping.)

	A	B ₁	B ₂	B ₃	B ₄	B ₅	B ₆
A	ϵ_A	t	t	t	t	t	t
B ₁	t	ϵ_B					
B ₂	t		ϵ_B				
B ₃	t			ϵ_B			
B ₄	t				ϵ_B		
B ₅	t					ϵ_B	
B ₆	t						ϵ_B

$= H$

2. The def'n of WF is

$$W(r-R) = \frac{1}{N} \sum_k e^{-ik \cdot R} \psi_k(r) = \frac{a}{2\pi} \int_{-\pi/a}^{\pi/a} e^{-ik \cdot R} \psi_k(r) dk$$

$$\text{Substituting: } W(r-R) = \frac{a}{2\pi} \int_{-\pi/a}^{\pi/a} dk e^{-ikR} \sum_{R'} e^{ikR'} \phi(r-R')$$

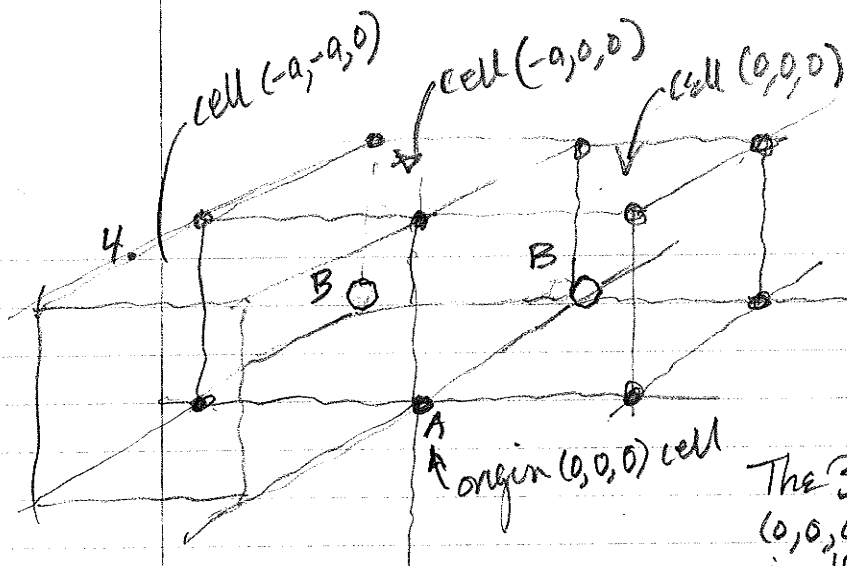
$$= \sum_{R'} \phi(r-R') \frac{a}{2\pi} \int_{-\pi/a}^{\pi/a} dk e^{-ik(R-R')} \quad \text{if } R \neq R'$$

$$\text{But } R=ma, R'=na \text{ for some integers } m, n \text{ so } \rightarrow e^{-i(m-n)\pi} - e^{i(m-n)\pi} = -2i \sin(m-n)\pi = 0.$$

$$\text{If } R'=R, \text{ the integral is } 2\pi/a. \text{ I.e. } \frac{a}{2\pi} \int_{-\pi/a}^{\pi/a} dk = \delta_{R,R'}$$

$$\text{Thus } \boxed{W(r-R) = \phi(r-R).}$$

3. There are several positives & negatives for each method.
Not try to list here.



Look at $H_{AB}(k) = \sum_R t_{AB}(R) e^{ik \cdot R}$ first.

What are the $\{R\}$ vectors for nearest neighbors to $(0,0,0)$?

The 3 unit cells pictured are $(0,0,0)$, $(-1,0,0)$, $(-1,-1,0)$, the other in this plane is $(0,-1,0)$.

The other 4 are lower by $-a\hat{z}$:
 $(0,0,-1)$, $(-1,0,-1)$, $(-1,-1,-1)$, $(0,-1,-1)$

t_{AB} 's are all the same.

Write in order of $R_z = 0$ and $R_z = -a$ pairs, it's systematic

$$t_{AB} \sum_R e^{ik \cdot R} = t_{AB} \left((1 + e^{-ik_z a}) + e^{-ik_x a} (1 + e^{-ik_z a}) + e^{-ik_y a} (1 + e^{-ik_z a}) + e^{-i(k_x + k_y)a} (1 + e^{-ik_z a}) \right)$$

$$= t_{AB} \left[(1 + e^{-ik_z a}) \left(1 + e^{-ik_x a} + e^{-ik_y a} + e^{-i(k_x + k_y)a} \right) \right]$$

Can be simplified somewhat but that is a correct result.

Now for $H_{AA}(k) = \sum_R t_{AA}(R) e^{ik \cdot R}$. This is easier, as all nearest neighbor A atoms lie in different cells $(\pm 1, 0, 0)$, $(0, \pm 1, 0)$, $(0, 0, \pm 1)$

Then $H_{AA}(k) = t_{AA} (2 \cos k_x a + 2 \cos k_y a + 2 \cos k_z a)$

$H_{BB}(k) = t_{BB} (\quad \quad \quad)$ by analogy.

$$H(k) = \begin{bmatrix} H_{AA}(k) & H_{AB}(k) \\ H_{AB}^*(k) & H_{BB}(k) \end{bmatrix}_0$$

5).

Number of Aluminum atom in one cube cell $\Rightarrow 4$.

Number of electron per atom $\Rightarrow 3$ either is OK

$$\text{electron density} = \frac{4 \times 3 \#}{(4 \text{ \AA})^3} = \frac{\sqrt[3]{3} \#}{16 \text{ \AA}^3} = \frac{\sqrt[3]{3} \#}{108.9 \text{ a.u.}^3} \text{ for } 1 \text{ a.u.} = 0.529 \text{ \AA}$$

$$\text{For } N = 2 \int \frac{d^3 k}{(2\pi)^3} = \frac{k_F^3}{3\pi^2} \Rightarrow k_F = (3\pi^2 n)^{1/3} = \left(3\pi^2 \frac{3}{16}\right)^{1/3} \text{ \AA}^{-1} \approx 6^{1/3} \text{ \AA}^{-1} \approx 1.8 \text{ \AA}^{-1}$$

$$\Rightarrow E_F = \frac{\hbar^2 k_F^2}{2m} = \frac{1}{2} (3\pi^2 n)^{2/3} \cdot \left(\frac{\hbar^2}{m a_0^3}\right) \cdot a_0^2 = \frac{1}{2} \left(3\pi^2 \cdot \frac{3}{108.9}\right)^{2/3} \cdot \frac{a_0^2}{a_0^3} \cdot H_a = 0.43 H_a.$$

$\hookrightarrow H_a$

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