## Tuning the two-dimensional electron gas at the LaAlO<sub>3</sub>/SrTiO<sub>3</sub>(001) interface by metallic contacts

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First principles calculations reveal that adding a metallic overlayer on LaAlO<sub>3</sub>/SrTiO<sub>3</sub>(001) eliminates the electric field within the polar LaAlO<sub>3</sub> film and thus suppresses the thickness-dependent insulator-to-metal transition observed in uncovered films. Independent of the LaAlO<sub>3</sub> thickness both the surface and the interface are metallic, with an enhanced interface carrier density relative to LaAlO<sub>3</sub>/SrTiO<sub>3</sub>(001) after the metallization transition. Moreover, a monolayer thick metallic Ti-contact exhibits a finite magnetic moment and for a thin SrTiO<sub>3</sub>-substrate induces a spin-polarized 2D electron gas at the *n*-type interface due to confinement effects. A diagram of band alignment in M/LaAlO<sub>3</sub>/SrTiO<sub>3</sub>(001) and Schottky barriers for M=Ti, Al, and Pt are provided.

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The (001) interface between the band insulators LaAlO<sub>3</sub> (LAO) and  $SrTiO_3$  (STO) provides remarkable examples of the novel functionalities that can arise at oxide interfaces, including a quasi two-dimensional electron gas (q2DEG)[1], superconductivity [2], magnetism [3] and even signatures of their coexistence [4, 5]. Moreover, a thickness-dependent transition from insulating to conducting behavior was reported in thin LaAlO<sub>3</sub> films on SrTiO<sub>3</sub>(001) at  $\sim 4$  monolayers (ML) LAO [6]. This insulator-metal transition (MIT) can be controlled reversibly via an electric field, e.g. by an atomic force microscopy (AFM) tip [7-9]. Recently, it was shown that an additional STO capping layer can trigger the MIT already at 2 ML of LAO, allowing stabilization of an electron-hole bilayer [10]. Density functional theory calculations (DFT) have demonstrated the emergence of an internal electric field for thin polar LAO overlayers[11-14]. Screening of the electric field by a strongly compensating lattice polarization in the LAO film[12, 15] allows several layers of LAO to remain insulating before an electronic reconstruction takes place at around 4-5 monolayers of LAO. Recent AFM experiments provide evidence for such an internal field in terms of a polarity-dependent asymmetry of the signal [16], but xray photoemission studies [17-19] have not been able to detect shifts or broadening of core-level spectra that would reflect an internal electric field. This discrepancy implies that besides the electronic reconstruction, extrinsic effects play a role, e.g. oxygen defects [20, 21], adsorbates such as water or hydrogen [22] or cation disorder [23, 24] (for detailed reviews on the experimental and theoretical work see [13, 25-27]).

The LAO/STO system is not only of fundamental scientific interest, but is also a promising candidate for the development of electronics and spintronics devices [28, 29]. For its incorporation in such devices the influence of metallic leads needs to be considered [30–32]. Metallic overlayers have been investigated on a variety of perovskite surfaces, such as SrTiO<sub>3</sub>(001) [33, 34], LaAlO<sub>3</sub>(001) [35, 36] or BaTiO<sub>3</sub>(001) [37, 38]. Furthermore, the magnetoelectric coupling between ferromagnetic Fe and Co films and ferroelectric BaTiO<sub>3</sub> and PbTiO<sub>3</sub>(001)-surfaces [39, 40] has been explored. However, the impact of a metallic overlayer on a buried oxide interface has not been addressed so far theoretically. Based on DFT calculations we show here that a metallic Ti overlayer not only acts as a Schottky barrier, but has a crucial influence on the electronic properties as it removes the internal electric field in the LAO film on STO(001). Despite the lack of a potential build up, the underlying STO layer is metallic and a significantly enhanced carrier concentration emerges at the *n*-type interface as compared to the system without metallic contact. Although bulk Ti is nonmagnetic, the undercoordinated Ti in the contact layer shows an enhanced tendency towards magnetism with a significant spin polarization and a magnetic moment of 0.60  $\mu_{\rm B}$ . Most interestingly, quantum confinement within the STO-substrate can induce spin-polarized carriers at the interface.

DFT calculations on nTi/mLAO/STO(001), where n(m)denotes the number of metallic overlayers (monolayers of LAO) were performed using the all-electron full-potential linearized augmented plane wave (FP-LAPW) method in the WIEN2k implementation [41] and the generalized gradient approximation (GGA)[42] of the exchange-correlation potential. Including an on-site Coulomb correction within the LDA/GGA+U approach [43] with U = 5 eV and J = 1 eVapplied on the Ti 3*d*-states and U = 7 eV on the La 4*f* states showed only small differences in electronic behavior. The calculations are carried out using a symmetric slab with LAO and Ti layers on both sides of the STO-substrate and a vacuum region between the periodic images of at least 10Å. The lateral lattice parameter is set to the GGA equilibrium lattice constant of STO (3.92 Å, slightly larger than the experimental value 3.905 Å) and the atomic positions are fully relaxed within tetragonal symmetry.

Fig. 1a shows the layer resolved density of states (LDOS) of a 1Ti/4LAO/STO(001) system, where the Ti atoms are adsorbed on top of the oxygen ions in the surface  $AIO_2$  layer. A striking feature is that the electric field of the uncovered LAO film (black line), expressed through an upward shift of the O 2p bands [12], is completely eliminated after the adsorption of the Ti overlayer. As a consequence of the van-

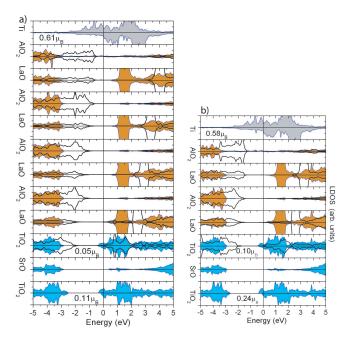


FIG. 1: (Color online) Layer resolved density of states (LDOS) of a) 1Ti/4LAO/STO(001) and b) 1Ti/2LAO/STO(001). The potential build up in the uncovered systems (black line) is canceled in the ones with a Ti overlayer (filled area). Additionally, there is a significant spin-polarization both in the surface Ti, as well as the interface TiO<sub>2</sub> layers in 1Ti/NLAO/STO(001).

ishing electric field within the LAO layer, no dependence of the electronic properties on the LAO thickness is expected, as opposed to the insulator-metal transition that occurs in the uncovered LAO/STO(001) system. Indeed the LDOS of 1Ti/2LAO/STO(001) (Fig. 1b, only 2 LAO layers) confirms a very similar behavior with no shifts of the O 2p bands in the LAO part. Despite the lack of a potential build up, there is a considerable occupation of the Ti 3d band at the interface and thus both the surface Ti layer and the interface are metallic. The charge transfer from the Ti adlayer allows the Ti+LAO+STO system to equilibrate in charge and potential, with the result being that the Fermi level lies just within the STO conduction band.

The low coordination of the surface Ti atoms enhances their tendency towards magnetism resulting in a magnetic moment of  $0.58\mu_{\rm B}$  in the surface layer. The electron gas at the interface is also spin-polarized with magnetic moments sensitive to the thickness of the LAO-spacer: for  $m_{LAO} = 4$  the magnetic moments are smaller ( $0.05/0.11\mu_{\rm B}$  in the interface (IF)/IF-1 layer) than for  $m_{LAO} = 2$  ( $0.10/0.24\mu_{\rm B}$  in IF/IF-1). Calculations performed within GGA+U for the latter case show a similar behavior but with an even stronger spin-polarization of carriers and magnetic moments of Ti of  $0.20/0.30 \mu_{\rm B}$  in the IF/IF-1 layer, respectively. Thus correlation corrections have only small influence on the overall band alignment, confirming that the observed behavior is not affected by the well known underestimation of band gaps of LDA/GGA.

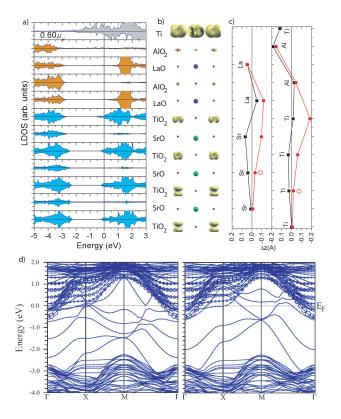


FIG. 2: (Color online) a) Layer resolved density of states (LDOS) of 1Ti/2LAO/STO(001) within GGA with a 6.5ML thick STOsubstrate; b) Side view of the system with the electron density integrated in the interval  $E_F - 0.5eV$  to  $E_F$ ; c) vertical displacements of cations and anions in 1Ti/2LAO/STO(001); d) majority and minority band structure where the 3*d* bands in the interface layer are emphasized with the  $d_{xy}$  orbital being the lowest lying level at  $\Gamma$ .

The calculations so far were performed with a rather thin substrate layer of 2.5 ML STO. To examine the dependence on the thickness of the substrate layer, we studied 1Ti/2LAO/STO(001) containing a 6.5 ML thick STO part. As shown in Fig. 2a, the most prominent difference to the system with a thin STO layer is the suppression of spin-polarization of carriers at the interface, indicating that the spin-polarization is a result of confinement effects in the thin STO layer. Apart from this, a notable band bending occurs in the STO part of the heterostructure. The largest occupation of the Ti 3d band arises at the interface, followed by a decreasing occupation in deeper layers. The electron density, integrated over states between  $E_F - 0.5 eV$  to  $E_F$  in Fig. 2b, reveals orbital polarization of the Ti electrons in the conduction band: predominantly  $d_{xy}$ character in the interface layer, nearly degenerate  $t_{2q}$  occupation in IF-1, and a preferential occupation of  $d_{xz}$ ,  $d_{yz}$  levels in the deeper layers. The band structure plotted in Fig. 2d shows that the conduction band minimum is at the  $\Gamma$ -point, formed by  $d_{xy}$  states of Ti in the interface layer. While the  $d_{xy}$  bands have a strong dispersion, the  $d_{xz}$ ,  $d_{yz}$  bands lie slightly higher in energy but are much heavier along the  $\Gamma - X$  direction. In addition to the orbital polarization of the filled bands, the

different band masses indicate a significant disparity in mobilities of electrons in the different  $t_{2g}$  orbitals. Similar multiple subband structure has been recently reported for LAO/STO superlattices [44, 45],  $\delta$ -doped LAO in STO [46] as well as doped STO(001)-surfaces [47, 48]. The remaining bands between -2.5eV and E<sub>F</sub> are associated with the surface Ti-layer.

The altered boundary conditions in 1Ti/LAO/STO(001) affect also the structural relaxations: In the uncapped LAO/STO(001) system a strong lattice polarization emerges primarily in the LAO film (with a buckling of  $\sim 0.26$ Å in the LaO and  $\sim 0.17$ Å in the subsurface AlO<sub>2</sub> layers) that opposes the internal electric field [12]. As seen in the LDOS plots, the formation of a chemical bond between the metallic layer and the LAO film cancels the electric field within LAO, and hence, as shown in Fig. 2c, cations and anions within the LAO film relax by similar amount with no appreciable buckling of the layers. On the other hand, the excess charge at the LAO/STO interface induces polarization in the STO substrate. The displacement between anions and cations is driven mainly by an outward oxygen shift and is largest at the interface (0.17Å in the interface  $TiO_2$  layer and 0.16Å in the next SrO layer) and decays in deeper layers away from the interface. This relaxation pattern resembles the one of n-type LAO/STO and LTO/STO superlattices [49, 50].

Increasing the thickness of Ti to 2 ML (with Ti in the second layer positioned above Al in the surface AlO<sub>2</sub>-layer and La in the subsurface LaO-layer) leads to a significant reduction of the spin-polarization of the Ti film: the magnetic moment is  $0.25\mu_{\rm B}$  in the surface and  $-0.10\mu_{\rm B}$  in the subsurface layer. This confirms that the high spin-polarization of Ti in 1Ti/mLAO/STO(001) is a result of the reduced coordination (for comparison, the magnetic moment of a free standing Ti layer is  $0.90\mu_{\rm B}$ ). The weaker binding to the oxide surface in 2Ti/2LAO/STO(001) is expressed in a longer Ti-O bond length of 2.06 Å compared to 2.00 Å in 1Ti/2LAO/STO(001). Despite these differences in the structural and magnetic properties of the metallic overlayer, the occupation of the Ti 3*d* band at the interface is very similar (cf. Fig.3a), but decreases quicker in the deeper layers.

Besides Ti, we have studied also Pt and Al overlayers and find that the overall behavior - strong reduction/cancellation of the electric field within LAO and metallization of the underlying STO - is robust for all studied metallic contacts. The layer resolved occupation of the Ti 3d band in SrTiO<sub>3</sub>(001) (Fig.3a) correlates with the positions of O1s core levels with respect to the Fermi level (cf. Fig.3b). The lowest O1s eigenvalue occurs at the interface, with the strongst binding energy for a Ti monolayer with a thin STO substrate (occupation of  $0.2e^{-}$  of the Ti 3d orbitals). For the thicker STO substrate, the ordering of 3d occupation is Al, Ti, 2Ti with relatively small differences. Finally, the lowest Ti 3d band occupation is for a Pt contact with the respective O1s eigenvalue at the interface being highest. These quantitative differences are associated with the different chemical bond between the metal overlayer and the surface AlO<sub>2</sub> layer: e.g. the bonding is strongly ionic in the case of Al with a significant charge transfer from Al

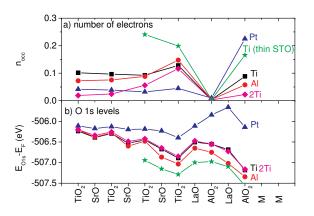


FIG. 3: a) Layer resolved electron occupation integrated between  $E_F - 0.65$  eV and  $E_F$  for 1ML Ti and thin (green stars)/thick STO substrate (black squares); 2ML Ti (magenta diamonds), as well as 1ML Pt (blue triangles) and 1ML Al (red circles). Occupation within SrTiO<sub>3</sub> layers is due to Ti 3*d* electrons, the one in the topmost AlO<sub>2</sub> arises from metal induced gap states (MIGS). Note that charge on the AlO<sub>2</sub> layer next to the interface is nearly vanishing in all cases, charge on the AO layers is zero (not shown). b) Positions of O1*s* states with respect to  $E_F$ .

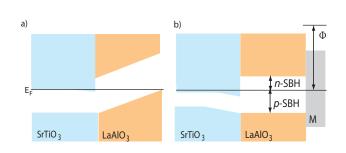


FIG. 4: Schematic band diagram of a) LAO/STO(001) at the verge of an electronic reconstruction at the critical LAO thickness; b) LAO/STO(001) covered by a metallic contact layer (M). Note, that the potential build up that leads to an electronic reconstruction in the uncovered LAO/STO(001) system at the critical thickness is strongly reduced/eliminated in M/LAO/STO(001).

to O and is much weaker for a Pt overlayer with its nearly filled 4d states. The bond strength variation is also reflected in a sizeable difference in bond distances ( $d_{A1-O} = 1.97$ Å,  $d_{Pt-O} = 2.31$ Å). Furthermore, in the systems with a thick STO-substrate the O1s levels in deeper layers away from the IF shift upwards and converge to a similar value indicating that the inner potential relaxes to the bulk STO value. The total 3d band occupation is similar for Ti (thin and thick STO slab) and A1 (~ 0.4e) and lower for 2Ti (0.22e) and Pt (0.16e).

The distinct mechanisms of formation of a q2DEG in LAO/STO(001) with and without a metallic contact are displayed in the schematic band diagram in Fig. 4. For

LAO/STO(001) a thickness dependent MIT occurs as a result of the potential buildup, where the electronic reconstruction involves the formation of holes at the surface and electrons at the interface. In contrast, for M/LAO/STO(001) the potential in LAO is flat regardless of the LAO or STO thickness. Simultaneously, a q2DEG with higher carrier density is formed at the interface. Only in the case of Pt (and possiblly other noble metals), likely due to the weaker bonding and smaller charge transfer to the oxide layer, a small residual slope within LAO is found, consistent with the recently measured potential build up in Pt/LAO/STO(001) [31].

The Schottky barrier height (SBH) between the metal and the oxide film is an important quantity which depends critically on the type of metal, the chemical bonding characteristics and the work function (see e.g. [34]): the *p*-SBH determined from the LDOS are 3.0 eV (Al), 2.8 eV (Ti) and 2.2 (Pt). The latter two values are slightly higher but show the same trend as recent results for Pt and Al on LAO(001) [36]. The corresponding *n*-SBH are 0.7 eV (Al), 0.9 eV (Ti) and 2.0 eV (Pt), determined from LDOS, or 1.6 eV higher in each case if the experimental band gap of LAO is considered. The conduction band offset (*n*-type SBH) is related to the effective work function of the system ( $\Phi^{AI} = 3.53 \text{ eV}$ ,  $\Phi^{Ti} = 4.05 \text{ eV}$ , and  $\Phi^{Pt} = 5.60 \text{ eV}$ ) and correlates also with the Ti 3*d* band occupation at the LAO/STO interface which is highest for an Al contact, followed by Ti and lowest for Pt.

These results show that metallic contacts ultimately change the electrostatic boundary conditions and represent a further powerful means besides oxygen defects[21] or adsorbates[22], oxide overlayers [10] or an external electric field [7–9] to tune the functionality at the LAO/STO(001) interface. Despite analogies to the adsorption of hydrogen on LAO/STO(001)[22], there are also subtle differences (*e.g.* the strong dependence of the potential slope on coverage in the latter system). These differences emphasize not only the importance of the electrostatic boundary conditions[51] but also of a precise understanding of structural effects and chemical bonding to LAO/STO(001) in order to achieve better understanding and control device performance.

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- [1] A. Ohtomo and H. Y. Hwang, Nature 427, 423 (2004).
- [2] N. Reyren et al., Science **317**, 1196 (2007).
- [3] A. Brinkmann et al., Nature Mater. 6, 493 (2007).
- [4] D. A. Dikin et al., arXiv:1103.4006v1 (2011).
- [5] L. Li et al., arXiv:1105.0235v1 (2011).
- [6] S. Thiel et al., Science 313, 1942 (2006).

- [7] C. Cen et al., Nature Mater. 7, 298 (2008).
- [8] F. Bi et al., Appl. Phys. Lett. 97, 173110 (2010).
- [9] Y. Z. Chen et al., Appl. Phys. Lett. 97, 123102 (2010).
- [10] R. Pentcheva et al., Phys. Rev. Lett. 104, 166804 (2010).
- [11] S. Ishibashi and K. Terakura, J. Phys. Soc. Jpn. 77, 104706 (2008).
- [12] R. Pentcheva and W. E. Pickett, Phys. Rev. Lett. **102**, 107602 (2009).
- [13] R. Pentcheva and W. E. Pickett, J. Phys.: Condens. Matter 22, 043001 (2010).
- [14] W.-J. Son, E. Cho, B. Lee, J. Lee, and S. Han, Phys. Rev. B 79, 245411 (2009).
- [15] S. A. Pauli et al., Phys. Rev. Lett. 106, 036101 (2011).
- [16] Y. Xie et al., Nano Lett. **10**, 2588 (2010).
- [17] Y. Segal et al., Phys. Rev. B 80, 241107(R) (2009).
- [18] M. Sing et al., Phys. Rev. Lett. 102, 176805 (2009).
- [19] S. A. Chambers et al., Surface Science Reports 65, 317 (2010).
- [20] Z. Zhong, P. X. Xu, and P. J. Kelly, Phys. Rev. B 82, 165127 (2010).
- [21] N. C. Bristowe, P. B. Littlewood, and E. Artacho, Phys. Rev. B 83, 205405 (2011).
- [22] W.-J. Son et al., J. Phys.: Condens. Matter 22, 315501 (2010).
- [23] P. R. Willmott et al., Phys. Rev. Lett. 99, 155502 (2007).
- [24] L. Qiao et al., J. Phys.: Condens. Matter 22, 312201 (2010).
- [25] M. Huijben et al., Adv. Mater. **21**, 1665 (2009).
- [26] H. Chen, A. M. Kolpak, and S. Ismail-Beigi, Adv. Mater. 22, 2881 (2010).
- [27] P. Zubko et al., Annu. Rev. Condens. Matter Phys. 2, 141 (2011).
- [28] J. Mannhart and D. G. Schlom, 327, 1607 (2010).
- [29] G. Cheng et al., Nature Nanotechnology 6, 343 (2011).
- [30] R. Jany et al., Appl. Phys. Lett. 96, 183504 (2010).
- [31] G. Singh-Bhalla et al., Nature Physics 7, 80 (2011).
- [32] Z. Liu et al., arXiv:1011.2629v1 (2010).
- [33] A. Asthagiri and D. S. Sholl, J. Chem. Phys. 116, 9914 (2002).
- [34] M. Mrovec et al., Phys. Rev. B 79, 245121 (2009).
- [35] A. Asthagiri and D. S. Sholl, Phys. Rev. B 73, 125432 (2006).
- [36] Y. F. Dong et al., Appl. Phys. Lett. 89, 122115 (2006).
- [37] N. Sai, A. M. Kolpak, and A. M. Rappe, Phys. Rev. B 72, 020101(R) (2005).
- [38] M. Stengel, D. Vanderbilt, and N. A. Spaldin, Phys. Rev. B 80, 224110 (2009).
- [39] C.-G. Duan, S. S. Jaswal, and E. Y. Tsymbal, Phys. Rev. Lett. 97, 047201 (2006).
- [40] M. Fechner et al., Phys. Rev. B 78, 212406 (2008).
- [41] P. Blaha et al., Wien2k, an augmented plane wave plus local orbitals program for calculating crystal properties, (Karlheinz Schwarz, Technische Universität Wien, Austria, 2001).
- [42] J. P. Perdew, K. Burke, and M. Ernzerhof, Phys. Rev. Lett. 77, 3865 (1996).
- [43] V. I. Anisimov et al., Phys. Rev. B 48, 16929 (1993).
- [44] Z. S. Popović, S. Satpathy, and R. M. Martin, Phys. Rev. Lett. 101, 256801 (2008).
- [45] K. Janicka, J. P. Velev, and E. Y. Tsymbal, Phys. Rev. Lett. 102, 106803 (2009).
- [46] P. V. Ong, J. Lee, and W. E. Pickett, Phys. Rev. B 83, 193106 (2011).
- [47] A. F. Santander-Syro et al., Nature 469, 189 (2011).
- [48] W. Meevasana et al., Nature Mater. 10, 114 (2011).
- [49] R. Pentcheva and W. E. Pickett, Phys. Rev. B. 78, 205106 (2008).
- [50] S. Okamoto, A. J. Millis, and N. A. Spaldin, Phys. Rev. Lett. 97, 056802 (2006).
- [51] M. Stengel, Phys. Rev. Lett. 106, 136803 (2011).